



Growth of CdS Nanoparticles to Fabricate Schottky Barrier

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Authors' contributions

This work was carried out in collaboration between all authors. All authors read and approved the final manuscript.

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ABSTRACT

CdS nanoparticles have been grown by a simple cost effective chemical reduction method and a Schottky barrier of gold/ nano CdS is fabricated. The grown nanoparticles are structurally characterized by transmission electron microscopy and x ray diffraction. The optical properties of nano CdS is characterized by optical absorption, photoluminescence study. The band gap of the CdS nanoparticles is increased as compared to CdS bulk form. Capacitance–voltage and current–voltage characteristics of gold / nano CdS Schottky barrier junction have been studied. It is found that these characteristics are influenced by surface or interface traps. The values of barrier height, ideality factor, donor concentration and series resistance are obtained from the reverse bias capacitance–voltage measurements.

Keywords: CdS nanoparticles; structural properties; optical properties; Au/n-CdS schottky barrier.

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1. INTRODUCTION

Semiconductor nanoparticles are promising material in electrical and optoelectronic devices. Properties of nanostructures such as structural, electrical, optical etc. are different from their bulk form due to mainly quantum confinement effect and surface to volume ratio [1-3]. Cadmium Sulphide a group II - group VI semiconductor having characteristic band gap 2.42 eV in bulk form has been used in different optoelectronic devices e.g. Solar cell, LED, Laser etc. [4-7]. CdS nanostructures based hetero junctions, Schottky barriers are important for application in such devices [8-10]. The Schottky barriers of CdS with different metals have been studied by researchers [11-13]. The electrical properties of Schottky devices are affected by metal semiconductor interface or surface properties of semiconductors [14]. Proper modification of surface states of semiconducting nanoparticles is still a challenge for researchers. In this work an effort has been made to grow CdS nanoparticles by a very cost effective and controlled way. The effect of nanoparticles surface on the formation of barrier is investigated to modify the Schottky device based on CdS nanoparticles. There are various physical and chemical methods to prepare CdS nanostructures [15-18]. We have followed a simple chemical reduction method to grow CdS nanoparticles which is cost effective also [19].

In this work CdS nanoparticles are prepared by a reliable low cost method to fabricate Schottky barrier with gold. The surface of the nanoparticles is modified by controlling the growth condition. The technique for preparation of nano CdS film on ITO coated glass is also cost effective. The structural and optical properties of synthesized CdS nanoparticles are characterized. The electrical properties of Au/n-CdS Schottky junction have been studied by current-voltage and capacitance – voltage measurements. The values of barrier height, ideality factor, and donor concentration are obtained by experiment results.

2. EXPERIMENTAL DETAILS

The CdS nanoparticles are synthesized by a chemical reduction method at room temperature. Cadmium chloride, sulphur powder and sodium borohydride are used to grow CdS nanoparticles. The structure of grown nanoparticles are characterized Transmission Electron Microscope JEOL JEM200 at 200 kV. Optical absorption of

the grown nanoparticles is performed by Shimadzu-Pharmaspec-1700 visible and ultraviolet spectrophotometer. Photoluminescence spectra of sample are observed by Perkin Elmer spectrophotometer. The procedure to grow CdS nanoparticles, structural, optical characterization of as prepared CdS nanoparticles is described elsewhere [19,20].

To fabricate Schottky junction a film of the CdS nanoparticles has been grown from the dispersed CdS nanoparticles on ITO coated glass. The pre-cleaned ITO coated glass substrate has been dipped in to the dispersed solution of CdS nanoparticles at least for 6 hrs. Uniformly thin film of CdS nanoparticles has been deposited on the glass substrate. Schottky junction is fabricated by evaporating gold (Au) dots of 2 mm diameter through a mask on CdS film. Fig. 1 shows the schematic diagram of fabricated Schottky barrier.

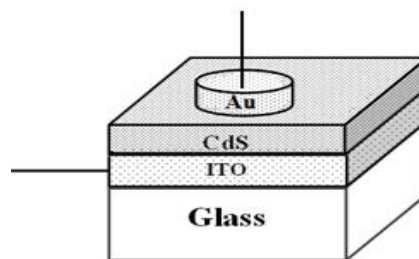


Fig. 1. Schematic diagram of fabricated Au/nanoCdS Schottky device

Current-voltage and capacitance-voltage measurements of Au/n-CdS Schottky junction are performed using HP4284A LCR meter and Keithley electrometer.

3. RESULTS AND DISCUSSION

The TEM image and corresponding selected area electron diffraction (SAD) pattern of grown CdS nanoparticles is shown in Fig. 2. The TEM image confirms that CdS nanoparticles are formed and agglomerated. The size of the as prepared nanoparticles is of the order of 11-14 nm. Patel et al prepared CdS nanoparticles of size 12 nm [14]. The SAD pattern displays the presence of diffraction rings which corresponds to the hexagonal wurtzite crystal phase of CdS.

Fig. 3 shows the x ray diffraction (XRD) pattern of the as prepared sample. The XRD pattern shows that synthesized nano CdS sample has hexagonal wurtzite structure [20]. The prominent

peaks shown in the XRD pattern are indexed with respective planes.

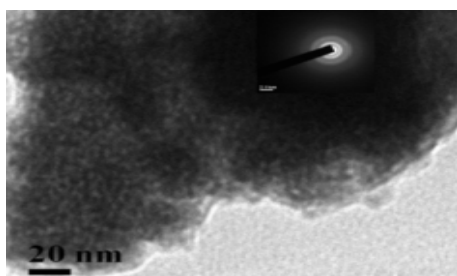


Fig. 2. TEM image and SAD pattern (inset) of as synthesized CdS nanoparticles

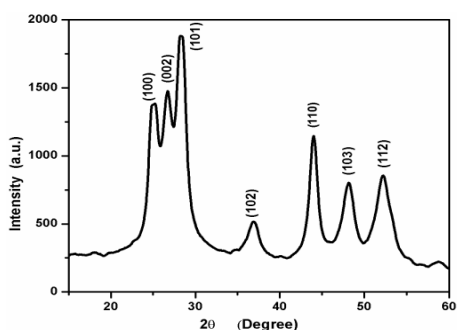


Fig. 3. The XRD pattern of the as grown CdS nanoparticles

The variation of optical absorbance of CdS nanoparticles with wavelength is shown in Fig. 4.

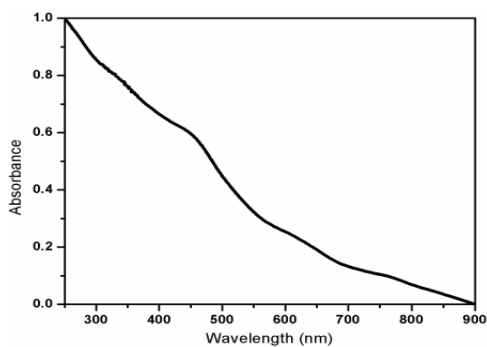


Fig. 4. The optical absorption spectrum of as prepared sample

The absorption spectrum is normalized. The band gap of the as-prepared nanoparticles is determined from the Tauc relation [21]

$$(ahv)^2 = C (hv - E_g) \quad (1)$$

Where C is a constant, E_g is the band gap of the semiconductor material and α is the absorption coefficient. Band gap of the CdS nanoparticles is calculated from $(ahv)^2$ vs. hv plot which is given in Fig. 5. The linear part of the curve is extrapolated to energy (hv) axis to determine band gap. The band gap is found to be 2.97 eV. Patidar et al. [16] obtained band gap of CdS nanoparticles on the order of 2.47-3.12 eV.

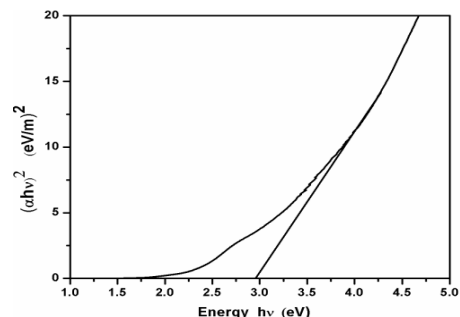


Fig. 5. The band gap determination curve for as prepared sample

The photoluminescence spectrum of as-prepared CdS sample is displayed in Fig. 6.

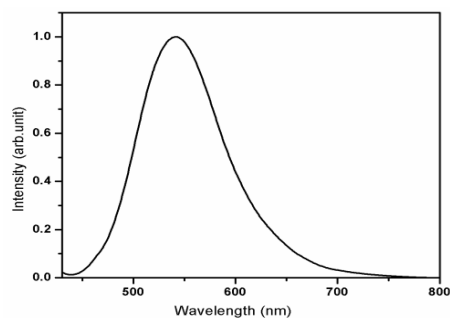


Fig. 6. The photoluminescence spectrum of as prepared sample

The photoluminescence intensity is normalized. Photoluminescence spectrum displays peak around 542 nm due to presence of surface states [22]. Wang et al. [19] observed photoluminescence peak of CdS nanoparticles at 560 nm.

3.1 Capacitance(C) –Voltage (V) Measurement

The C-V measurement of Au/n-CdS Schottky junction with reverse and forward biasing voltages at temperature 303 K is shown in Fig. 7.

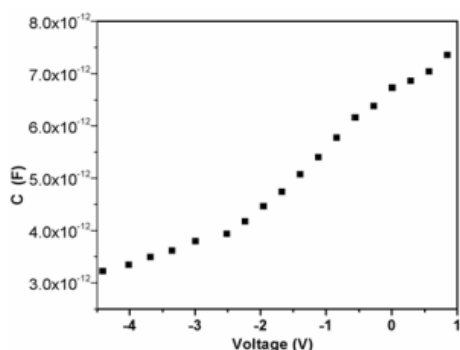


Fig. 7. The C-V characteristics of Au /nano CdS Schottky junction

The $1/C^2$ vs. V plot is given in Fig. 8. The carrier concentration, built-in-voltage is determined from the slope [23] and the intercept on the V axis of $1/C^2$ vs. V plot using the Mott-Schottky relation (2)

$$C^{-2} = \frac{2(V_b + V)}{q\epsilon\epsilon_0 A^2 N_d} \quad (2)$$

where N_d is the donor concentration, V_b is the built-in potential, q is the electronic charge, ϵ_0 is the permittivity of free space, ϵ is the dielectric constant of the semiconductor. W is the width of the depletion region. A is the area of the device. In Mott-Schottky relation it is assumed that surface or interface traps are absent, no interfacial layer is present between metal and semiconductor [14].

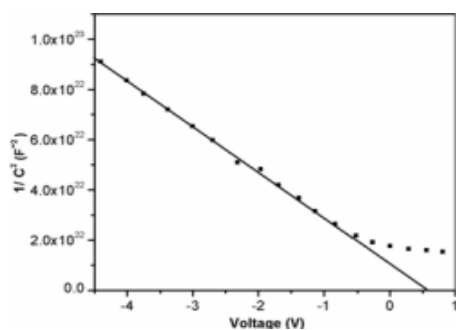


Fig. 8. Reverse bias $1/C^2$ vs V characteristics of Au/n-CdS Schottky barrier

The obtained values of N_d and V_b are given in Table 1. The value of barrier height ϕ_b is calculated by the following relation

$$\Phi_b = V_b + V_p \quad (3)$$

Where V_p is the potential difference between the Fermi level and the top of the valance band in CdS. V_p is calculated by knowing the donar concentration N_d and value of N_c is obtained from the following relation

$$V_p = KT \ln (N_c / N_d) \quad (4)$$

Table 1. Different parameters of Au/n-CdS Schottky junction at temperature 303K

V_b (V)	N_d (cm^{-3})	Φ (eV)	η
0.56	5.41×10^{15}	0.80	2.19

Where $N_c = 1.5 \times 10^{20} \text{ cm}^{-3}$ is the density of states in the conduction band for CdS [13]. The calculated barrier height value for the Au/n-CdS Schottky junction is given in Table 1. Farag et al. [13] obtained barrier height 0.76-0.86 eV. Patel et al. [14] obtained barrier height of 0.82 eV. It is seen from the result that C-V characteristics of Au/n-CdS Schottky junction is influenced by surface traps [24,25].

3.2 Current (I)–voltage (V) Characteristics

The I-V characteristics of the Au/n-CdS device under forward and reverse biasing conditions at 303K is shown in Fig. 9.

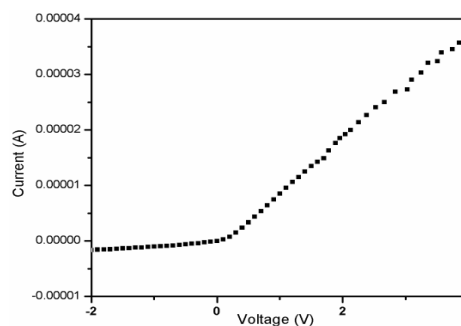


Fig. 9. The I-V characteristics of the Au/n-CdS device in forward and reverse biasing condition

The electron affinity of n-type CdS is 4.8 eV [26] while the work function of gold (Au) is about 5.25 eV [27]. So a Schottky barrier should be formed at the contact interfaces of Au/n-CdS. For a metal semiconductor Schottky barrier diode assuming thermionic emission to be the dominant transport mechanism the relationship between current and voltage is given by

$$I = I_s [\exp(qV/\eta kT) - 1] \quad (5)$$

Where I_s is saturation current, V is applied voltage, η is the ideality factor, k is the Boltzmann constant, T is the absolute temperature in Kelvin. I_s is described as

$$I_s = AA^*T^2 \exp(-q\Phi/kT) \quad (6)$$

Where A is the area of device, A^* is the modified Richardson constant and Φ is the effective barrier height from metal to semiconductor.

The saturation current is determined from a plot of $\ln(I)$ vs voltage (V), where I_s is obtained as the intercept of the linear region of the $\ln(I)$ vs V curve extrapolated to zero voltage. Fig. 10 shows the current voltage characteristics $\ln(I)$ vs V plot. The saturation current is evaluated 1.42×10^{-7} A.

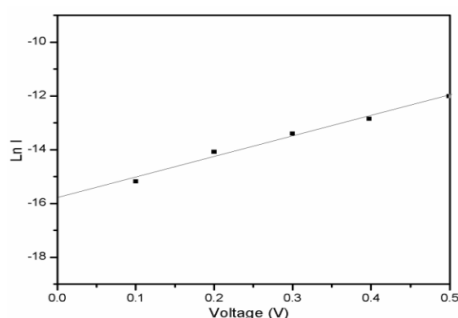


Fig. 10. Current-voltage characteristics of the Au/nano-CdS Schottky barrier plotted as $\ln(I) - V$

Taking into consideration of series resistance R_s the equation can be expressed as

$$I = I_s [\exp(q(V - IR_s)/\eta kT) - 1] \quad (7)$$

The equation can be differentiated as [28]

$$dV/d(\ln I) = IR_s + \eta kT/q \quad (8)$$

The plot associated with Eq. (8) $dV/d(\ln I)$ vs I is given in Fig. 11.

The series resistance R_s is calculated from the slope of $dV/d(\ln I)$ versus I characteristic according to equation (7) [29]. The series resistance R_s here includes the contact resistance. While η is evaluated from the $dV/d(\ln I)$ axis intercept of the line fit shown in Fig. 9. The series resistance is found to be 8.27 k Ω . The ideality factor in the room temperature is listed in Table 1.

The obtained high series resistance of the Au/n-CdS Schottky device may be attributed to the

high resistance of the starting CdS material or to the interfacial layer created between the metal and CdS [4]. The ideality factor is determined to be 2.19. Which is greater than typical value between 1 to 2 [30]. But the value of ideality factor greater than 2 is also possible [31]. Patel et al found ideality factor of Au/n CdS Schottky barrier 1.8, 6.0 [14]. Ideality factor greater than 2 has been obtained with Schottky devices made of nanostructures. An oxide layer may be present between semiconductor and metal [32]. The high value of η may be due to large recombination within the interfacial layer [15,33] which exists mainly in the semiconductor side.

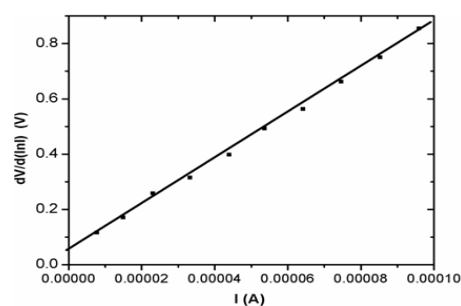


Fig. 11. $dV/d(\ln I)$ versus I plot for Au/n-CdS device

4. CONCLUSION

CdS nanoparticles are synthesized by a cost effective chemical method. The structural and optical characterizations of the synthesized CdS nanoparticles have been performed. We have also fabricated Au/ n- CdS Schottky junction with the grown CdS nanoparticles. The C-V and I-V characteristics of the Au/n-CdS Schottky junction have been studied. The values of built in potential, saturation current, barrier height, ideality factor, series resistance, the density of interface states have been calculated. It is found that the I-V and C-V characteristics of the Au/n-CdS Schottky junction are influenced by the surface states or interface traps.

COMPETING INTERESTS

Authors have declared that no competing interests exist.

REFERENCES

1. Peng XG, Manna L, Yang WD, Wickham J, Scher E, Kadavanich A, Alivisatos AP. Shape control of CdSe nanocrystals. Nature. 2000;404:59.

2. Trindade T, Brien PO', Pickett NL. Nanocrystalline semiconductors: synthesis, properties, and perspectives. *Chem. Mater.* 2001;13(11):38-43.
3. Bawendi MG, Carroll PJ, Wilson WL, Brus LE. Luminescence properties of cadmium selenide quantum crystallites: Resonance between interior and surface localized states. *J. Chem. Phys.* 1992;96:946.
4. Pal R, Bhattacharya D, Maity AB, Choudhury S, Pal AK. Nanostructured ZnTe films prepared by D.C. magnetron sputtering. *Nanostruct. Mater.* 1994;4:329.
5. Lin YF, Hsu YJ, Lu SY, Chen KT, Tseng TY. Well-Aligned Ternary Cd_{1-x}Zn_xS Nanowire Arrays and Their Composition-Dependent Field Emission Properties. *J. Phys. Chem. C.* 2007;111:13418.
6. Law M, Green LE, Johnson JC, Saykally R, Yang P. Nanowire dye-sensitized solar cells. *Nat. Mater.* 2005;4:455.
7. Kang Y, Kim D. Well-aligned CdS nanoro / conjugated polymer solar cells. *Sol. Energy Mater. Sol. Cells* 2006;90:166.
8. Ye Y, Dai L, Wu P C, Liu C, Sun T, Qin GG. Schottky junction photovoltaic devices based on CdS single nanobelts. *Nanotechnology.* 2009;20:375202.
9. Mandal SK, Maity AB, Dutta J, Chaudhuri S, Pal AK. Au/CdS Schottky Diode Fabricated with Nanocrystalline CdS Layer. *Phys. Stat. Sol. (a)* 1997;163(2): 433.
10. Lin YF, Song JH, Ding Y, Lu SY, Wang Z L. Alternating the Output of a CdS Nanowire Nanogenerator by a White-Light-Stimulated Optoelectronic Effect. *Adv. Mater.* 2008;20:3127.
11. Chand S, Kumar J. Effects of barrier height distribution on the behavior of a Schottky diode. *J. Appl. Phys.* 1997;82:5005.
12. Didden A, Battjes H, Machunze R, Dam B, Krol R. Titanium nitride: A new Ohmic contact material for n-type CdS. *J. Appl. Phys.* 2011;110:033717.
13. Farag AAM, Yahia IS, Fadel M. Electrical and photovoltaic characteristics of Al/n-CdS Schottky diode. *Int. J. Hydrogen Energy.* 2009;34(11):4906.
14. Patel BK, Nanda KK, Sahu SN. Interface characterization of nanocrystalline CdS/Au junction by current-voltage and capacitance-voltage studies. *J. Appl. Phys.* 1999;85(7):366.
15. Chaure NB, Bordas S, Samantilleke AP, Chaure SN, Haig J, Dharmadasa IM. Investigation of electronic quality of chemical bath deposited cadmium sulphide layers used in thin film photovoltaic solar cells. *Thin Solid Films.* 2003;437:10.
16. Patidar D, Rathore KS, Saxena NS, Sharma K, Sharma TP. Energy band gap studies of CdS nanomaterials. *J. Nano. Res.* 2008;3:97.
17. Xu D, Xu Y, Chen D, Guo G, Gui L, Tang Y. Preparation and characterization of CdS nanowire arrays by dc electrodeposit in porous anodic aluminum oxide templates. *Chem. Phys. Letters.* 2000;325:340.
18. Shen XP, Yuan AH, Wang F, Hong JM, Xu Z. Fabrication of well-aligned CdS nanotubes by CVD-template method. *Solid State Commun.* 2005;133:19.
19. Wang W, Germanenko I, El-Shall MS. Room-temperature synthesis and characterization of nanocrystalline CdS, ZnS, and Cd_xZn_{1-x}S. *Chem. Matter.* 2002; 14:3028.
20. Bhattacharya R, Das TK, Saha S. Synthesis and characterization of CdS nanoparticles. *J Mater Sci: Mater. Electron* 2011;22:1761.
21. Ghobadi N. Band gap determination using absorption spectrum fitting procedure. *Int. Nano Lett.* 2013;3:2.
22. Okamoto S, Kanemitsu Y, Hosokawa H, Murakoshi K, Yanagida S. Size selective photoluminescence excitation spectroscopy in CdSe nanocrystals. *Solid State Commun.* 1998;105(1):7
23. Sze SM. *Physics of semiconductor devices*, 2nd ed (John Wiley & Sons, New York. 1981;249.
24. Sahu SN, Chandra S. Chemical-bath-deposited CdS and CdS: Li films and their use in photo electrochemical solar cells. *Sol. Cells.* 1987;22:163.
25. Zhang ZY, Jin CH, Liang X L, Chen Q, Peng LM. Current-voltage characteristics and parameter retrieval of semiconducting nanowires. *Appl. Phys. Lett.* 2006; 88:073102.
26. Swank RK. Surface Properties of II-VI Compounds. *Phys. Rev.* 1967;153:844.
27. Rusu PC, Brocks G. Surface dipoles and work functions of Alkylthiolates and fluorinated Alkylthiolates on Au(111). *J. Phys. Chem. B.* 2006 10: 22628.
28. Cheung SK, Cheung NW. Extraction of Schottky diode parameters from forward current-voltage characteristics. *Appl. Phys. Lett.* 1986;49:85.

29. Quan DT, Hbib H. High barrier height Au/n-type InP Schottky contacts with a POxNyHz interfacial layer. Solid-State Electron. 1993;36:339.
30. Verschraegen J, Burgelman M, Penndorf J. Temperature dependence of the diode ideality factor in CuInS₂-on-Cu-tape solar cells. Thin Solid Films. 2005;480-481:307.
31. Maruska HP, Namavar F, Kalkhoran NH. Current injection mechanism for porous-silicon transparent surface light-emitting diodes. Appl. Phys. Lett. 1992;61:1338.
32. Maruska HP, Ghosh AK, Eustace DJ, Feng T. Current injection mechanism for porous-silicon transparent surface light-emitting diodes. J. Appl. Phys. 1983;49:3490.
33. Lao CS, Liu J, Gao PX, Zhang LY, Davidovic D, Tummala R, Wang ZL. ZnO Nanobelt / Nanowire Schottky Diodes Formed by Dielectrophoresis Alignment across Au Electrodes Nano Lett. 2006;2: 263.

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